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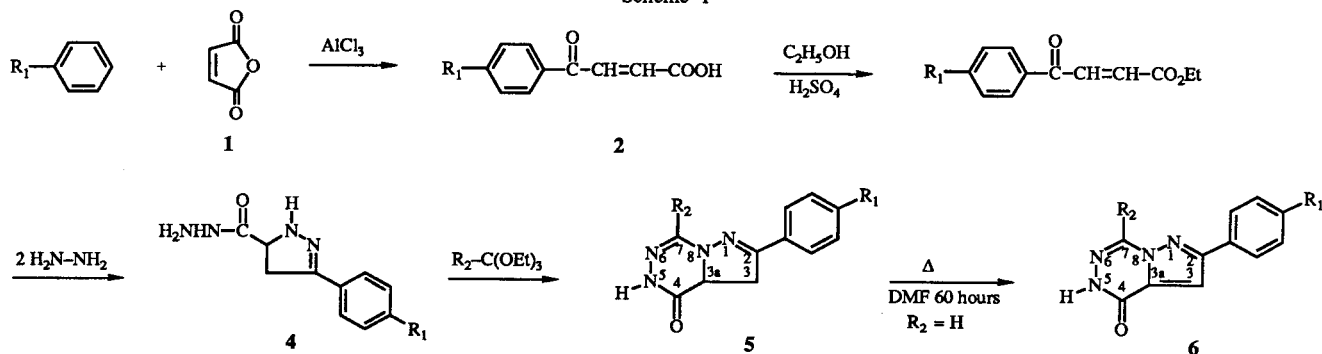
The preparation of new dihydropyrazolo[1,5-d][1,2,4]triazinones involved the formation of carboxylic acid hydrazides *via* the appropriate ethyl aroylacrylates, followed by condensation with an orthoester. The synthesis of some derivatives substituted on the nitrogen atom in the 5-position of the triazine ring was reported and their anticonvulsant activity was evaluated.

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Fused heterocyclic systems containing a triazine ring were largely investigated because they were effective in many pharmacological areas. These derivatives presented a great number of biological properties especially antipressant [1], antitumor [2], antiparasitic [3], antifungal [4], bactericidal [5], antiallergic [6] and antiinflammatory effects [7].

Many authors have proposed various procedures to obtain pyrazolo[1,5-d][1,2,4]triazines [8-15] but no pyrazolo-triazin-4-ones hydrogenated in the 3 and 3a-positions and bearing an aryl substituent on the pyrazoline ring were described. We report in this paper a simple synthesis of this new bicyclic ring system by the methods shown in Scheme 1. 3-Aroylacrylic acids **2** were prepared by Friedel-

Scheme 1



Scheme 2

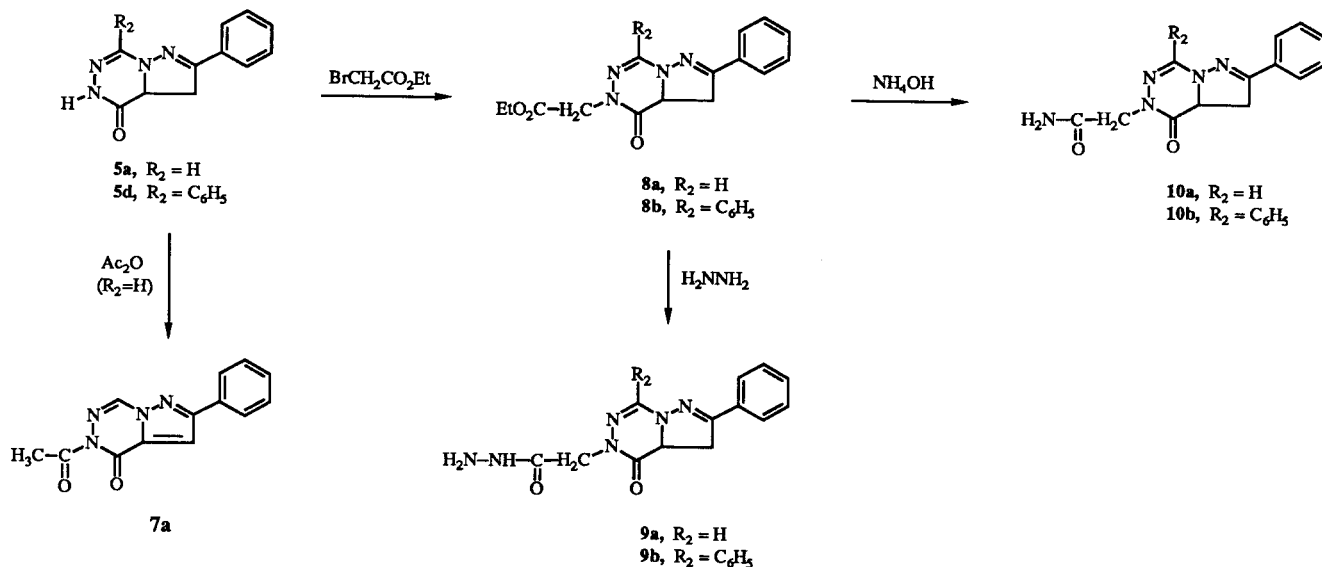


Table 1  
Physical and analytical data for Compounds 5,6,7,8,9 and 10

Compound No.	R <sub>1</sub>	R <sub>2</sub>	Mp (C°)	Yield %	Molecular Formula (MW)	Analysis			
						C	H	N	Cl
5a	H	H	215	68	C <sub>11</sub> H <sub>10</sub> N <sub>4</sub> O (214)	61.68	4.67	26.17	
						61.64	4.53	26.28	
5b	H	CH <sub>3</sub>	236	77	C <sub>12</sub> H <sub>12</sub> N <sub>4</sub> O (228)	63.16	5.26	24.56	
						62.95	5.33	24.41	
5c	H	C <sub>2</sub> H <sub>5</sub>	177	67	C <sub>13</sub> H <sub>14</sub> N <sub>4</sub> O (242)	64.46	5.79	23.14	
						64.53	5.63	23.02	
5d	H	C <sub>6</sub> H <sub>5</sub>	210	45	C <sub>17</sub> H <sub>14</sub> N <sub>4</sub> O (290)	70.34	4.83	19.31	
						70.35	4.90	19.23	
5e	OCH <sub>3</sub>	H	220	67	C <sub>12</sub> H <sub>12</sub> N <sub>4</sub> O <sub>2</sub> (244)	59.02	4.92	22.95	
						58.83	4.82	22.88	
5f	OCH <sub>3</sub>	CH <sub>3</sub>	204	88	C <sub>13</sub> H <sub>14</sub> N <sub>4</sub> O <sub>2</sub> (258)	60.46	5.43	21.71	
						60.39	5.28	21.81	
5g	OCH <sub>3</sub>	C <sub>2</sub> H <sub>5</sub>	214	66	C <sub>14</sub> H <sub>16</sub> N <sub>4</sub> O <sub>2</sub> (272)	61.76	5.88	20.59	
						61.76	5.93	20.42	
5h	OCH <sub>3</sub>	C <sub>6</sub> H <sub>5</sub>	190	47	C <sub>18</sub> H <sub>16</sub> N <sub>4</sub> O <sub>2</sub> (320)	67.50	5.00	17.50	
						67.64	4.92	17.40	
5i	Cl	H	258	80	C <sub>11</sub> H <sub>9</sub> N <sub>4</sub> ClO (248.5)	53.12	3.62	22.54	14.28
						53.14	3.69	22.75	14.10
5j	Cl	CH <sub>3</sub>	228	45	C <sub>12</sub> H <sub>11</sub> N <sub>4</sub> ClO (262.5)	54.86	4.19	21.33	13.52
						55.06	3.92	21.32	13.81
5k	Cl	C <sub>2</sub> H <sub>5</sub>	195	43	C <sub>13</sub> H <sub>13</sub> N <sub>4</sub> ClO (276.5)	56.42	4.70	20.25	12.84
						56.33	4.72	19.94	13.02
5l	Cl	C <sub>6</sub> H <sub>5</sub>	225	15	C <sub>17</sub> H <sub>13</sub> N <sub>4</sub> ClO (324.5)	62.86	4.01	17.26	10.94
						62.83	3.95	17.15	11.07
6a	H	H	263	78	C <sub>11</sub> H <sub>8</sub> N <sub>4</sub> O (212)	62.26	3.77	26.42	
						62.15	3.69	26.30	
6b	OCH <sub>3</sub>	H	287	66	C <sub>12</sub> H <sub>10</sub> N <sub>4</sub> O <sub>2</sub> (242)	59.50	4.13	23.14	
						59.40	4.04	23.29	
6c	Cl	H	255	60	C <sub>11</sub> H <sub>7</sub> N <sub>4</sub> ClO (246.5)	53.55	2.84	22.72	14.40
						53.61	2.97	22.68	14.33
7a	H	H	262	27	C <sub>13</sub> H <sub>10</sub> N <sub>2</sub> N <sub>4</sub> O <sub>2</sub> (254)	61.42	3.93	22.05	
						61.31	3.86	21.91	
8a	H	H	108	56	C <sub>15</sub> N <sub>16</sub> N <sub>4</sub> O <sub>3</sub> (300)	60.00	5.33	18.67	
						59.95	5.27	18.56	
8b	H	C <sub>6</sub> H <sub>5</sub>	176	97	C <sub>21</sub> H <sub>20</sub> N <sub>4</sub> O <sub>3</sub> (376)	67.02	5.32	14.89	
						67.12	5.13	14.86	
9a	H	H	154	33	C <sub>13</sub> H <sub>14</sub> N <sub>6</sub> O <sub>2</sub> (286)	54.55	4.89	29.37	
						54.68	4.80	29.32	
9b	H	C <sub>6</sub> H <sub>5</sub>	255	64	C <sub>19</sub> H <sub>18</sub> N <sub>6</sub> O <sub>2</sub> (362)	62.98	4.97	23.21	
						62.80	4.98	23.34	
10a	H	H	150	76	C <sub>13</sub> H <sub>13</sub> N <sub>5</sub> O <sub>2</sub> ·2H <sub>2</sub> O (307)	50.81	5.54	22.80	
						50.75	5.68	22.87	
10b	H	C <sub>6</sub> H <sub>5</sub>	220	36	C <sub>19</sub> H <sub>17</sub> N <sub>5</sub> O <sub>2</sub> (347)	65.71	4.90	20.17	
						65.62	4.85	20.29	

Crafts acylation using maleic anhydride [16]. After esterification of acrylic acids, the resulting derivatives **3** were treated with hydrazine hydrate to produce 4,5-dihydropyrazole carboxylic acid hydrazides **4**. Condensation of compounds **4** with orthoesters afforded the expected dihydropyrazolotriazinones **5**. On continued heating in DMF for 60 hours, compounds **5** were dehydrogenated in the 3 and 3a-positions and led to pyrazolotriazinones **6**.

The analogy of these compounds with *N*-substituted pyridazinones that exhibited anticonvulsant properties [17] prompted us to prepare some pyrazolotriazinone derivatives substituted in the 5-position by acetyl, hydrazide and amide chains (Scheme 2). Acetylation of compound **5a** was performed with acetic anhydride; simultaneous dehydrogenation of the pyrazoline nucleus occurred under reflux and acetyl derivative **7a** was obtained. Treatment of

Table 2  
IR and <sup>1</sup>H-NMR for Compounds 5, 6, 7, 8, 9 and 10

Compound No.	IR (KBr) $\nu$ (cm <sup>-1</sup> )			<sup>1</sup> H-NMR (DMSO-d <sub>6</sub> ) $\delta$ (ppm)
	NH	C=O	C=N C=C	
5a	3200	1660	1640, 1610 1580, 1460	3.6 (m, 2H, CH <sub>2</sub> ), 4.8 (m, 1H, CH), 7.6 (m, 5H, C <sub>6</sub> H <sub>5</sub> ), 7.8 (s, 1H, =CH), 10.7 (br s, 1H, NH)
5b	3240	1670	1610, 1580 1455	2.2 (s, 3H, CH <sub>3</sub> ), 3.5 (m, 2H, CH <sub>2</sub> ), 4.7 (m, 1H, CH), 7.7 (m, 5H, C <sub>6</sub> H <sub>5</sub> ), 10.4 (br s, 1H, NH)
5c	3200	1660	1600, 1580 1450	1.3 (t, 3H, CH <sub>3</sub> ), 2.7 (q, 2H, CH <sub>2</sub> ), 3.6 (m, 2H, CH <sub>2</sub> ), 4.7 (m, 1H, CH), 7.7 (m, 5H, C <sub>6</sub> H <sub>5</sub> ), 10.4 (br s, 1H, NH)
5d	3220	1660	1605, 1590 1500, 1450	3.8 (m, 2H, CH <sub>2</sub> ), 5.0 (m, 1H, CH), 7.7 (m, 10H, 2C <sub>6</sub> H <sub>5</sub> ), 11.2 (br s, 1H, NH)
5e	3220	1675	1610, 1580 1510, 1460	3.6 (m, 2H, CH <sub>2</sub> ), 3.8 (s, 3H, OCH <sub>3</sub> ), 4.7 (m, 1H, CH), 7.4 (m, 4H, C <sub>6</sub> H <sub>4</sub> ), 7.8 (s, 1H, =CH), 10.6 (br s, 1H, NH)
5f	3260	1680	1600, 1510 1470	2.3 (s, 3H, CH <sub>3</sub> ), 3.5 (m, 2H, CH <sub>2</sub> ), (m, 2H, CH <sub>2</sub> ), 3.9 (s, 3H, OCH <sub>3</sub> ), 4.7 (m, 1H, CH), 7.5 (m, 4H, C <sub>6</sub> H <sub>4</sub> ), 10.0 (br s, 1H, NH)
5g	3240	1670	1610, 1580 1510, 1460	1.3 (t, 3H, CH <sub>3</sub> ), 2.8 (q, 2H, CH <sub>2</sub> ), 3.6 (m, 2H, CH <sub>2</sub> ), 4.0 (s, 3H, OCH <sub>3</sub> ), 4.7 (m, 1H, CH), 7.5 (m, 4H, C <sub>6</sub> H <sub>4</sub> ), 10.6 (br s, 1H, NH)
5h	3260	1670	1610, 1570 1440	3.8 (m, 2H, CH <sub>2</sub> ), 3.9 (s, 3H, OCH <sub>3</sub> ), 5.1 (m, 1H, CH), 7.5 (m, 9H, C <sub>6</sub> H <sub>5</sub> , C <sub>6</sub> H <sub>4</sub> ), 11.1 (br s, 1H, NH)
5i	3200	1660	1620, 1580 1470	3.5 (m, 2H, CH <sub>2</sub> ), 4.8 (m, 1H, CH), 7.6 (m, 4H, C <sub>6</sub> H <sub>4</sub> ), 7.8 (s, 1H, =CH), 10.7 (br s, 1H, NH)
5j	3220	1670	1610, 1580 1470	2.2 (s, 3H, CH <sub>3</sub> ), 3.5 (m, 2H, CH <sub>2</sub> ), 4.7 (m, 1H, CH), 7.6 (m, 4H, C <sub>6</sub> H <sub>4</sub> ), 10.3 (br s, 1H, NH)
5k	3260	1680	1620, 1580 1450	1.2 (t, 3H, CH <sub>3</sub> ), 3.4 (q, 2H, CH <sub>2</sub> ), 3.5 (m, 2H, CH <sub>2</sub> ), 4.8 (m, 2H, CH <sub>2</sub> ), 7.7 (m, 4H, C <sub>6</sub> H <sub>4</sub> ), 10.6 (br s, 1H, NH)
5l	3300	1660	1590, 1570 1480	3.6 (m, 2H, CH <sub>2</sub> ), 5.0 (m, 1H, CH), 7.6 (m, 9H, C <sub>6</sub> H <sub>5</sub> , C <sub>6</sub> H <sub>4</sub> ), 11.2 (br s, 1H, NH)
6a	3200	1660	1610, 1550 1450	7.7 (m, 5H, C <sub>6</sub> H <sub>5</sub> ), 7.8 (s, 1H, =C <sub>3</sub> H), 9.1 (s, 1H, =C <sub>7</sub> H), 12.6 (br s, 1H, NH)
6b	3200	1660	1600, 1560 1430	3.9 (s, 3H, OCH <sub>3</sub> ), 7.7 (m, 4H, C <sub>6</sub> H <sub>4</sub> ), 7.9 (s, 1H, =C <sub>3</sub> H), 9.2 (s, 1H, =C <sub>7</sub> H), 12.3 (br s, 1H, NH)
6c	3200	1670	1600, 1500 1450	7.8 (m, 4H, C <sub>6</sub> H <sub>4</sub> ), 8.2 (s, 1H, =C <sub>3</sub> H), 9.1 (s, 1H, =C <sub>7</sub> H), 13.1 (br s, 1H, NH)
7a	-	1765	1630, 1550 1690 1470	2.5 (s, 3H, CH <sub>3</sub> ), 7.9 (m, 5H, C <sub>6</sub> H <sub>5</sub> ), 8.0 (s, 1H, =C <sub>3</sub> H), 9.2 (s, 1H, =C <sub>7</sub> H)
8a	-	1740	1615, 1600 1445	1.2 (t, 3H, CH <sub>3</sub> ), 3.4 (m, 2H, CH <sub>2</sub> ), 4.1 (q, 2H, OCH <sub>2</sub> ), 4.3 (s, 2H, NCH <sub>2</sub> ), 4.8 (m, 1H, CH), 7.5 (m, 5H, C <sub>6</sub> H <sub>5</sub> ), 7.9 (s, 1H, =CH)
8b	-	1750	1600, 1550 1450	1.3 (t, 3H, CH <sub>3</sub> ), 3.8 (m, 2H, CH <sub>2</sub> ), 4.2 (q, 2H, OCH <sub>2</sub> ), 4.7 (s, 2H, NCH <sub>2</sub> ), 5.1 (m, 1H, CH), 7.9 (m, 10H, 2C <sub>6</sub> H <sub>5</sub> )
9a	3300	1660	1590, 1500 1625 1450	3.3 (m, 2H, CH <sub>2</sub> ), 3.9 (br s, 2H, NH <sub>2</sub> ), 4.2 (s, 2H, NCH <sub>2</sub> ), 4.3 (m, 1H, CH), 7.5 (m, 6H, C <sub>6</sub> H <sub>5</sub> , =CH), 9.3 (br s, 1H, NH)
9b	3310	1660	1590, 1500 1625 1450	3.9 (m, 2H, CH <sub>2</sub> ), 4.3 (br s, 2H, NH <sub>2</sub> ), 4.4 (s, 2H, NCH <sub>2</sub> ), 5.2 (m, 1H, CH), 7.7 (m, 10H, 2C <sub>6</sub> H <sub>5</sub> ), 9.4 (br s, 1H, NH)
10a	3300	1690	1600, 1450 1660	3.6 (m, 2H, CH <sub>2</sub> ), 3.9 (s, 2H, NCH <sub>2</sub> ), 4.2 (m, 1H, CH), 7.6 (m, 6H, C <sub>6</sub> H <sub>5</sub> , =CH), 8.0 (br s, 6H, NH <sub>2</sub> , 2H <sub>2</sub> O)
10b	3410	1690	1600, 1500 1650 1450	3.9 (m, 2H, CH <sub>2</sub> ), 4.4 (s, 2H, NCH <sub>2</sub> ), 5.1 (m, 1H, CH), 6.9 (br s, 2H, NH <sub>2</sub> ), 7.9 (m, 10H, 2C <sub>6</sub> H <sub>5</sub> )

5a and 5d with ethyl bromoacetate gave the corresponding 5-substituted pyrazolotriazinones 8a and 8b, which were hydrolysed by hydrazine hydrate or aqueous ammonia to give hydrazides 9a-b or amides 10a-b, respectively.

The structure of compounds 5, 6, 7, 8, 9 and 10 was established from their analytical and spectral data (Tables

1, 2, 3). In Table 3 we only reported selected data of the <sup>13</sup>C nmr spectra.

Derivatives 7, 9 and 10 were evaluated for anticonvulsant activity against electrically induced seizures. In the maximal electroshock test [18], the only pyrazolotriazinones which displayed any noteworthy anticonvulsant effects, were 9a and 10b. At 10 mg/kg orally, both com-

pounds protected against seizures in 42% of the animals.

Table 3

<sup>13</sup>C-NMR for Compounds 5 and 6

Compound No.	C <sub>3</sub>	C <sub>3a</sub>	C <sub>7</sub>	C <sub>2</sub>	C <sub>4</sub>
5a	35.6	56.5	135.3	153.4	162.3
5b	35.4	56.6	143.1	152.5	162.6
5c	35.1	56.7	147.2	152.4	162.8
5d	36.0	56.3	144.3	151.7	163.8
5e	35.8	56.4	135.4	153.2	162.2
5f	35.5	56.6	144.1	154.1	162.3
5g	35.4	56.6	147.1	152.4	162.9
5h	36.1	56.1	144.4	151.6	163.7
5i	35.6	56.7	135.2	152.4	162.3
5j	35.3	56.7	143.1	151.7	162.7
5k	35.2	56.9	147.1	151.5	162.9
5l	36.0	56.5	144.2	150.8	163.7
6a	102.6	135.4	130.3	154.3	154.4
6b	102.0	135.4	130.2	154.3	154.4
6c	102.7	135.5	130.2	153.2	154.2

### EXPERIMENTAL

All melting points were determined on a Reichert apparatus and are uncorrected. The infrared spectra were recorded on a Beckman 4240 spectrophotometer. The proton nmr spectra were recorded on a Varian EM 360 A in DMSO-d<sub>6</sub>. Resonance positions are given on the δ scale (parts per million) relative to internal tetramethylsilane. The nmr peaks were designated as follows: s, singlet; br s, broad singlet; t, triplet; q, quadruplet; m, multiplet. The <sup>13</sup>C nmr spectra were recorded on a Jeol FX 60 in DMSO-d<sub>6</sub>. Elemental analysis were performed at the Service Central d'Analyses, Centre National de la Recherche Scientifique, 69390 Vernaison, France. Esters **3** were prepared from acryloylacrylic acids using the method of Delaby [19].

#### 3-Aryl-4,5-dihydropyrazole-5-carboxylic Acid Hydrazides **4**.

Pyrazolecarboxylic acid hydrazides **4** were synthesized using literature procedures [19,20].

#### 2-Aryl-3,3a-dihydro-4-oxo-5H-pyrazolo[1,5-d][1,2,4]triazines and 7-substituted Derivatives **5a-l**.

A solution of 0.05 mole of compound **4** and 0.05 mole of the appropriate orthoester in 30 ml of DMF was refluxed for 4 hours. In most cases, the crude products precipitated after cooling and were filtered; otherwise the solvent was evaporated under reduced pressure and the residual mixture triturated with ethyl ether. Crude products were recrystallized from DMF (**5a-c**, **5e-g**, **5i**) or ethanol (**5d**, **5h**, **5j-l**).

#### 2-Aryl-4-oxo-5H-pyrazolo[1,5-d][1,2,4]triazines **6a-c**.

A solution of 0.05 mole of the appropriate compound **5** in 30 ml of DMF was refluxed for 60 hours. After evaporation *in vacuo*, the crude product was washed with ethyl ether and recrystallized from methanol.

#### 5-Acetyl-2-phenyl-4-oxopyrazolo[1,5-d][1,2,4]triazine **7a**.

A solution of 2.14 g (0.01 mole) of pyrazolo[1,5-d][1,2,4]triazin-4(5H)-one **5a** in 30 ml of acetic anhydride was heated under reflux for 1 hour. Then the solution was cooled and the crude product which separated was filtered off and recrystallized from ethanol.

#### 2-Phenyl-3,3a-dihydro-4-oxopyrazolo[1,5-d][1,2,4]triazin-5-ylacetic Acid Ethyl Ester **8a** and 7-Phenyl Derivative **8b**.

A mixture of 0.01 mole of the compound **5a** or **5d**, 2.5 g (0.015 mole) of ethyl bromoacetate and 2.0 g (0.015 mole) of anhydrous potassium carbonate in acetone (100 ml) was refluxed under stirring for 24 hours in an oil-bath. The mixture was filtered hot, the filtrate was evaporated to dryness *in vacuo*, and the residue was triturated with diisopropyl ether. Compounds **8a** and **8b** were recrystallized from ethanol.

#### [2-Phenyl-3,3a-dihydro-4-oxopyrazolo[1,5-d][1,2,4]triazin-5-yl]-acetohydrazide **9a** and 7-Phenyl Derivative **9b**.

A mixture of the ester **8a** or **8b** (0.01 mole) in 20 ml of hydrazine hydrate and 20 ml of ethanol was refluxed for 4 hours. Then the mixture was cooled and the crude product which separated was filtered off and recrystallized from ethanol.

#### [2-Phenyl-3,3a-dihydro-4-oxopyrazolo[1,5-d][1,2,4]triazin-5-yl]-acetamide **10a** and 7-Phenyl Derivative **10b**.

A suspension of 0.005 mole of ester **8a** or **8b** in concentrated aqueous ammonia (15 ml) and ethanol (5 ml) was heated in a bomb apparatus at 100° for 24 hours. After cooling, the solvent was evaporated to dryness *in vacuo*. The solid residue was triturated with ethyl ether, until crystallization. Compounds **10a-b** were recrystallized from ethanol.

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